

Research Article

# Development, validation, and accreditation of a method for the determination of <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb in Cephalopods using inductively coupled plasma mass spectrometry (ICP-MS)



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#### **Abstract**

A sensitive, accurate and precise method was developed and validated for the trace level determination of  $^{75}$ As,  $^{111}$ Cd,  $^{201}$ Hg, and  $^{208}$ Pb in Cephalopods using inductively coupled plasma mass spectrometry. The selectivity and linearity, limit of detection (LOD) and limit of quantification (LOQ), recovery study, repeatability and within-laboratory reproducibility, Horwitz ratio (HORRAT), measurement uncertainty and fitness of purpose of the method were determined. The LOD for  $^{75}$ As,  $^{111}$ Cd,  $^{201}$ Hg, and  $^{208}$ Pb were found to be 14.68, 12.70, 5.13, and 2.74 µg/kg respectively whereas, the LOQ obtained were 48.94, 42.33, 17.11 and 9.13 µg/kg respectively. The relative standard deviation for the repeatability and within-laboratory reproducibility for all determined trace elements were below 3 and 16%, respectively. The HORRATr and HORRAT<sub>R</sub> values of repeatability and reproducibility were lower than 2. Furthermore, the recovery values of the spiked samples ranged from 81.79 to 101.67%. These results complied with the performance criteria established by Council Regulation (EC) 333/2007 and the clause 5.4.5.2, ISO/IEC 17025:2005. Therefore, the procedure is implemented for the routine analysis to determine trace level quantification of  $^{75}$ As,  $^{111}$ Cd,  $^{201}$ Hg, and  $^{208}$ Pb in Cephalopods.

 $\textbf{Keywords} \ \ \textbf{Validation process} \cdot \textbf{Limit of detection (LOD)} \cdot \textbf{Limit of quantification (LOQ)} \cdot \textbf{Trace elements} \cdot \textbf{Cephalopods} \cdot \textbf{ICP-MS} \cdot \textbf{Microwave digestion}$ 

# 1 Introduction

Trace elements are present in every ecosystem in the world [1]. These elements can be soluble in water and react with organic matter forming complexes and chelates, which increase its solubility, availability, and dispersal [2, 3]. The metal species can be mobilized from the environment, and accumulated in the biota in water, and are subsequently transferred to humans through the food chain [4]. Metals, such as arsenic, mercury, lead, and cadmium are toxic, even in trace amounts [5]. Though, the increase of these metals and metalloids in marine systems has become a risk to human health due to their toxic effects [6]. To protect

the public health at present, many countries enforce regulations on contamination level of toxic metals in fish and fish products. However, only three metals, cadmium, mercury, and, lead have been included in the regulations of the European Union for hazardous metals [7], while the Food Safety and Standards Authority of India [8] and United States Food and Drug Administration (USFDA) has included arsenic in the list [9]. These regulations require precise and accurate analytical method for the determination of heavy metals contaminant. For this reason, there is a growing global interest in performing validation procedures of the analytical method for legislation

<sup>☐</sup> Gupta Prakash Chandra, prakashgupta5@gmail.com | <sup>1</sup>Export Inspection Agency- Mumbai, E-3 MIDC, Andheri (East), Mumbai, Maharashtra 400093, India. <sup>2</sup>Export Inspection Council, 3rd Floor, NDYMCA Cultural Centre Building, 1 Jai Singh Road, New Delhi 110001, India



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implementation to show that the method produces reliable results and provide accurate and reproducible results.

A number of analytical techniques have been used for validating method for monitoring heavy metals in fish and fish products like flame atomic absorption spectroscopy (FAAS), graphite furnace atomic absorption spectroscopy (GFAAS), inductively coupled plasma-atomic emission spectrometry (ICP-AES) but inductively coupled plasma mass spectrometry (ICP-MS) is being the most sophisticated and reliable technique [10-13]. However, it is also an important point of consideration that the presence of spectral interferences (undesirable signals interfering with the element signal) and non-spectral interferences (physical and chemical interferences, also called matrix interferences) may result in false-positive analytical results [14]. Therefore, in order to eliminate or control effects of these interferences, it is necessary to use an optimized and validated ICP-MS method which permits to obtain the quantification of elements with the accuracy and precision. However, to the best of our knowledge, no accredited methods have been reported in the literature concerning the simultaneous determination of <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb in Cephalopods samples. The aim of the current study was to develop a rapid, low-cost analysis, accurate, and precise method for the determination of <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb in Cephalopods by inductively coupled plasma mass spectrometry (ICP-MS) according to the Council Regulation (EC) 333/2007 [16] and clause 5.4.5.2, ISO/IEC 17025: 2005 [15].

#### 2 Materials and methods

#### 2.1 Instrumentation

A standard Agilent 7800 ICP-MS, which includes High Matrix Introduction (HMI) system with SPS 4 autosampler, was used for the analysis. The ICP-MS was configured with the standard sample introduction system consisting of a micromist glass concentric nebulizer, a standard double-pass peltier-cooled spray chamber cooled at 2 °C, quartz spray chamber, and quartz torch with 2.5 mm id injector. Interface consisted of a nickel-plated copper sampling cone and a nickel skimmer cone. The operating conditions of instrument mentioned in Table 1 which has been selected on the basis to minimize interferences.

Tuning of the instrument was done by using 1 ppb (<sup>7</sup>Li, <sup>24</sup>Mg, <sup>59</sup>Co, <sup>89</sup>Y, <sup>140</sup>Ce, <sup>205</sup>TI) tuning solution in normal mode (no-gas mode) and helium mode to ensured optimal performances in terms of sensitivity, oxide ratio, and doubly charged ratio and to minimize interferences. In normal mode acquired element were <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb whereas, <sup>75</sup>As was acquired in helium (He) collision

Table 1 ICP-MS operating conditions

Parameter	Value
RF power (W)	1550
Sampling depth (mm)	10
Carrier gas (L/min)	1
Helium gas (mL/min)	4.3
Energy discrimination (V)	5
Nebulizer pump speed (rps)	0.1
Nebulizer carrier gas flow (L/min)	1
Lens extract 1(V)	0
Lens extract 2 (V)	-200
Omega bias (V)	-80
Omega lens (V)	9.5
Cell entrance (V)	-40
Cell exit (V)	-60
Deflect (V)	1.8
Plate bias (V)	-55
OctP bias (V)	-18
OctP RF (V)	200
Monitored elements	<sup>75</sup> As, <sup>111</sup> Cd, <sup>201</sup> Hg, <sup>208</sup> Pb

mode. The digestions were performed on a multiwave Go (Anton Paar).

#### 2.2 Reagents and chemicals

Ultrapure water (18.2 M $\Omega$  cm at 25 °C) from a Millipore Direct-Q-UV water purifier (Millipore) was used to prepare all standards and samples. Hydrogen peroxide (30%) (Merck India), Nitric acid (69.0–70.0%) and ACS Reagent grade hydrochloric acid (36.5–38.0%) were of JT Baker. The purity of argon plasma gas and helium collision gas were 99.999%.

# 2.3 Sample collection, preparation, and digestion

The Cephalopods samples were collected from the fish market, Andheri West, Mumbai, Maharashtra 400053, India and transported to the laboratory in the refrigerated condition. Samples were milled and homogenized in a homogenizer (Deimos, Astori Tecnica, Italy) as they arrived in the laboratory with a stainless steel cutter to obtain a representative homogenous subsample. The homogenized samples were labeled with a unique code and stored at  $-20\pm2$  °C till analysis. All necessary precaution has been taken to avoid secondary contaminants while preparing homogeneous samples.

The digestion of samples was performed as following 5 ml of concentrated  $HNO_3$ , 1 mL of HCl and 1.0 ml

Table 2 Parameters for microwave digestion

Step	Ramp time (Min)	Temp (°C)	Hold time (Min)
1	10	100	10
2	15	180	15

of concentrated  $\rm H_2O_2$  were added to 1.0 g of test sample accurately weighed in the digestion vessel. After a few minutes, the vessels were closed and then placed in the microwave digestion system. The digestion was performed as per the temperature control programme summarized in Table 2.

The digested samples were left to cool, quantitatively transferred to 50 ml volumetric flasks, diluted to the mark with Milli-Q water and stored in a refrigerator prior to ICPMS analysis, which was performed within 24 h after the digestion. All of the decomposed sample solutions were clear, without visible residues. The digestion vessels were cleaned after each run using the same reagents and program that was applied for sample digestion, followed by rinsing with Milli-Q water.

# 2.4 Range of the linearity and calibration curve

National Institute of Standards and Technology (NIST) traceable standards solutions of As, Cd, Hg, and Pb, 1000 mg/l in  $2\% \text{ HNO}_3$  were procured from Merck India. The standard blank solution was prepared using  $5\% \text{ HNO}_3$  and 0.5% HCl. The calibration range selected from 50 to 150% in relation to the expected result of analysis. The linearity of the calibration curve considered acceptable when the coefficient of regression ( $r^2$ ) was at levels equal to at least 0.99.

## 2.5 Experimental set up of the method validation

Different experiments were set up in order to investigate the performance characteristics of the method in terms of specificity, linearity, limit of detection (LOD) and limit of quantification (LOQ), measuring range, repeatability, within-laboratory reproducibility, recovery, as per Commission Regulation 333/2007/EC [16]. The method validation was performed by spiking blank matrix at 0.5, 1.0 and 1.5 times the maximum level as per Commission Regulation (EC) 1881/2006 for cadmium, lead, and mercury [17]. However, for arsenic 0.5, 1.0 and 1.5 times of 1 ppm was considered. The dilution factor 50 was considered for the calculation of analyte concentration. In order to check cross contaminations during the validation study procedural blank was prepared along with the samples.

The analytical sensitivity of ICP-MS was assessed by determining the values of limit of detection (LOD) and limit of quantification (LOQ). The values of LOD and LOQ were calculated as 3.3 SD/b and 10SD/b respectively, where SD is the standard deviation of the intercept and b is the slope of the obtained calibration curve. According to Commission Regulation (EC) 333/2007, the LOD should be equal to three-tenths of LOQ. Whereas, LOQ for Hg, Cd, and Pb should be less than one-fifth of the maximum level.

To check the accuracy of the analytical method, the recovery studies were performed in order to confirm no loss or contamination occurred during sample preparation and matrix interferences throughout the measurement step. Certified reference materials or control samples were not available, thus, the recovery studies were performed by spiking with the known amount of concentration prepared from NIST traceable standard in the blank sample according to the Table 3. The recovery was calculated as recovery % = 100C/spiked concentration where C is the element concentration found.

#### 2.6 Precision

The analytical precision was determined by assessing the repeatability and within-laboratory reproducibility of instrument response to analyte according to the Commission Regulation (EC) 657/2002 [18]. To assess the repeatability the analysis was done for three different concentrations levels according to Table 3, and in each level, the analysis was performed with six replicates. To assess the within-laboratory reproducibility the steps were repeated on two other occasions by a different analyst. The repeatability and within-laboratory reproducibility were calculated as the relative standard deviation.

The acceptance of repeatability and within-lab reproducibility Horwitz approach described in EURACHEM [19] and Thompson et al. [20] was used. This approach involves the Horwitz equation to calculate the relative standard deviation and the value was used in HORRAT ratio which was compared with acceptance values to finally evaluate acceptance of repeatability and within-lab-reproducibility.

The Horwitz equation used to calculate predicted relative standard deviation is,  $PRSD_R = 2C^{-0.15}$ , where, C is the

Table 3 Concentration levels for the analytes

Element	Level 1 (μg/Kg)	Level 2 (μg/Kg)	Level 3 (μg/Kg)
<sup>75</sup> As	500	1000	1500
<sup>111</sup> Cd	500	1000	1500
<sup>201</sup> Hg	250	500	750
<sup>208</sup> Pb	150	300	450

concentration ratio. Mean concentration, standard deviation, and relative standard deviations were calculated from results generated under repeatability (RSD<sub>r</sub>) and reproducibility (RSD<sub>R</sub>) conditions. HORRAT ratio was computed from the data of calculated RSD<sub>r</sub>, RSD<sub>R</sub>, PRSD<sub>r</sub> and PRSD<sub>R</sub> as HORRATr = RSDr (%)/PRSD<sub>r</sub> (%) and HORRAT<sub>R</sub> = RSD<sub>R</sub> (%)/PRSD<sub>R</sub> (%). According to Commission Regulation (EC) 836/2011 [16], the HORRATr and HORRAT<sub>R</sub> values should be less than two.

## 2.7 Measurement uncertainty

The combined standard measurement uncertainty ( $\rm U_c$ ) was calculated based on the Eurachem/Citac Guidelines [19], from the summary squared of several independent parameters such as (a) the mass uncertainty; (b) the dilution volume uncertainty; (c) the calibration uncertainty; (d) the bias uncertainty, as estimated by the recovery tests; and (e) the precision uncertainty, as estimated by the RSD<sub>R</sub>% values for the three different concentration levels under reproducibility conditions. The choice of factor k is 2 based on the confidence level of 95%.

Methods suitable for official control must produce results with combined standard measurement uncertainties (U<sub>c</sub>) less than the maximum standard measurement uncertainty (U<sub>f</sub>) calculated using the formula:  $U_f = \sqrt{(LOD/2)^2 + (\propto \mathbb{C})^2} \text{ where, LOD is the limit of detection of the method (µg/kg), C is the concentration of interest (µg/kg), <math display="inline">\alpha$  is a numeric factor to be used depending on the value of C [16].

#### 3 Result and discussion

In this study, an ICP-MS method for quantitative analysis of, <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb in Cephalopods was developed and validated.

# 3.1 Selectivity and linearity

Before, proceeding with the validation procedure the analytical method selectivity towards naturally present substances (metabolites, endogenous substances, matrix constituents, etc.) has ensured. To verify the selectivity calibration curves using aqueous standard solutions and matrix matching calibration curves were compared. The result showed that slope of the calibration curve from aqueous standards is almost similar to the slope of the curve plotted by matrix matching calibration curves indicating no matrix interferences. Therefore, aqueous calibration curves were used for calibration in all the experiments.

Table 4 shows the equation of a straight line and linear regression coefficient (r<sup>2</sup>) of a calibration curve (Fig. 1) of

Table 4 Linear regression data

Parameters	<sup>75</sup> As	<sup>111</sup> Cd	<sup>201</sup> Hg	<sup>208</sup> Pb
Linear range	1–50 μg/L	1-50 μg/L	0.1-20 μg/L	0.1–50 μg/L
r <sup>2</sup>	0.9977	0.9997	0.9997	0.9986
Slope	3763.4119	9370.0905	2248.3088	29761.8986
Intercept	904.1167	253.6167	40.0433	10549.7433

the determined elements. Results indicated that the linear regression model was acceptable for the selected analyte in the defined range with a satisfactory correlation coefficient ( $r^2 \ge 0.99$ ).

# 3.2 Limit of detection (LOD) and limit of quantification (LOQ)

The LOD for  $^{75}$ As,  $^{111}$ Cd,  $^{201}$ Hg, and  $^{208}$ Pb was found to be 14.68, 12.70, 5.13, and 2.74 µg/kg, whereas, LOQ obtained was 48.94, 42.33, 17.11 and 9.13 µg/kg respectively (Table 5). The value of LOD and LOQ comply with the Commission Regulation (EC) 333/2007, laying down the methods of sampling and analysis for the official control of the levels of trace elements and processing contaminants in foodstuffs. Furthermore, measuring range was found from 10 to 2500 µg/kg, which shows that developed method is sensitive and can quantify elements at trace level for the assessment of Cephalopods.

#### 3.3 Recovery study

The recovery values for accuracy studies spiked with different levels of  $^{75}$ As,  $^{111}$ Cd,  $^{201}$ Hg, and  $^{208}$ Pb are presented in Table 6. In the present study, the mean recovery ranged between 81.79-101.67% at all three spiked levels. The obtained results are within  $\pm$  20% of the target value thus the developed method was considered as "fit for purpose".

# 3.4 Repeatability and within-laboratory reproducibility

The data in Table 6 shows the repeatability, within-laboratory reproducibility expressed with the relative standard deviation and HORRAT. The RSD<sub>r</sub> % values obtained between 0.55 and 2.99 however, RSD<sub>R</sub> % values ranging between 1.83 and 15.86. According to Horwitz, as cited from Gonzalez and Herrador [21] the maximum RSD value acceptable for the analyte at the level of 100  $\mu$ g/kg and 1 mg/kg is 22.6 and 16% respectively. The values of HORRAT<sub>r</sub> and HORRAT<sub>R</sub> for repeatability and reproducibility was lower than 2 at all the concentration levels which in accordance to the performance criteria for methods of analysis as per Commission Regulation (EC) 836/2011

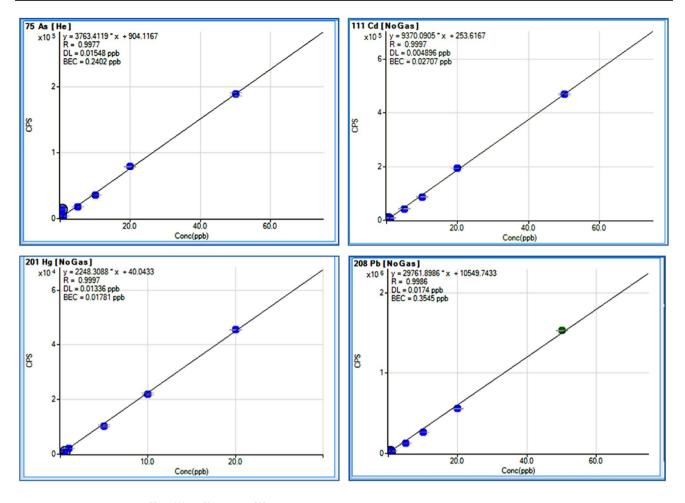


Fig. 1 Calibration curves for <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb

**Table 5** Limit of detection, quantification limits and measuring range

Element	LOD (µg/Kg)	LOQ (µg/Kg)	Measuring range (μg/ Kg)
<sup>75</sup> As	14.68	48.94	50-2500
<sup>111</sup> Cd	12.70	42.33	50-2500
<sup>201</sup> Hg	5.13	17.11	18-2500
<sup>208</sup> Pb	2.74	9.13	10-2500

[16]. Therefore, it can be stated that the developed method exhibited a good precision.

#### 3.5 Quality control programs

Internal quality control charts (IQCs) were prepared to monitor whether results are reliable enough to be reported. The goal of IQCs is the extension of method validation continuously checking the accuracy of analytical data obtained from routine analysis in the laboratory.

The analytical system is under control if no more than 5% of the measured values exceed the warning limits and none of them the action or control limits [22]. The external quality assurance program was maintained through the participation in proficiency programs organized by Export Inspection Agency -Kolkata, India. The results of proficiency test were within  $\pm 2$  Z scores.

# 3.6 Measurement uncertainty and fitness for purpose

The data of combined standard measurement uncertainty ( $U_c$ ) and maximum standard measurement uncertainty ( $U_f$ ) are summarized in Table 7. The results show that the method is suitable for the official control of <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb in Cephalopods samples because it produced results with combined standard measurement uncertainties less than the maximum standard measurement uncertainty.

Table 6 Recoveries, repeatability and within-laboratory reproducibility, and HORRAT ratio of the method

Analyte	Spiked analyte concen- tration (µg/kg)	Calculated analyte concentration (µg/kg)	Recovery (%)	RSD <sub>r</sub> %	RSD <sub>R</sub> %	HORRAT <sub>r</sub>	HORRAT <sub>R</sub>
<sup>75</sup> As	500	455.00	90.99	1.33	6.30	0.08	0.22
	1000	903.30	90.33	0.55	15.48	0.03	0.55
	1500	1343.75	89.58	1.37	3.51	0.09	0.16
<sup>111</sup> Cd	500	508.35	101.67	1.87	1.83	0.11	0.10
	1000	992.13	99.21	2.66	15.86	0.17	0.58
	1500	1447.80	96.52	2.18	2.71	0.15	0.16
<sup>201</sup> Hg	250	228.76	91.50	2.12	5.11	0.11	0.18
	500	437.05	87.41	2.99	13.21	0.16	0.46
	750	675.35	90.05	1.90	2.47	0.11	0.13
<sup>208</sup> Pb	150	132.42	88.28	2.80	3.60	0.13	0.15
	300	245.38	81.79	1.70	13.53	0.09	0.40
	450	377.90	83.98	2.29	3.01	0.13	0.15

**Table 7** Combined standard measurement uncertainty and maximum standard measurement uncertainty

Analyte	Spiked analyte concentration (µg/kg)	Combined measurement uncertainty (U <sub>c</sub> ), μg/kg	Maximum standard measurement uncertainty ( $U_f$ ), $\mu g/kg$
<sup>75</sup> As	1000	89.00	150.18
<sup>111</sup> Cd	1000	39.00	150.13
<sup>201</sup> Hg	500	15.00	75.04
<sup>208</sup> Pb	300	9.00	54.02

#### 4 Conclusions

The current work describes development and validation of a method for the trace level analysis of <sup>75</sup>As, <sup>111</sup>Cd, <sup>201</sup>Hg, and <sup>208</sup>Pb in Cephalopods as per Council Regulation (EC) 333/2007 and ISO/IEC 17025:2005 using inductively coupled plasma mass spectrometry (ICP-MS). The validation results have been presented and organized in the tables to provide an overview of the method's performance. The proposed method has met all performance criteria for a method of analysis defined in 333/2007/EC for LOD, LOQ, accuracy, precision, and considered as "fit for purpose". The method was accredited and implemented in the routine analysis and will also be helpful for monitoring trace elements for ensuring compliance in Cephalopods.

# **Compliance with ethical standards**

**Conflict of interest** On behalf of all authors, the corresponding author states that there is no conflict of interest.

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